First highly stereoselective synthesis of (+)-dihydrosesamin, a trisubstituted tetrahydrofuran-type of lignan, by using highly *erythro*-selective aldol condensation

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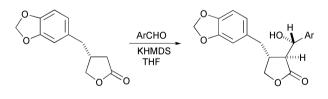
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A 2,3,4-tri-substituted tetrahydrofuran-type lignan, (+)-dihydrosesamin **2**, was stereoselectively synthesized by using *erythro*-selective aldol condensation of the potassium enolate from (3R)-3-(3,4-methylenedioxyphenyl)-4-butanolide **3** with piperonal as a key reaction. This is the first stereoselective synthesis of (+)-dihydrosesamin, which is the enantiomer of the natural product.

Introduction

A 2,3,4-trisubstituted tetrahydrofuran lignan, (-)-dihydrosesamin 1,¹ has been isolated from *Daphne tangutica* which had been used as an abortifacient and a medicine for rheumatism and toothache. Though the biological activity of (-)-dihydrosesamin 1 is unknown, tetrahydrofuran lignans are known to be antioxidants, PAF inhibitors and stress compounds in plants.² Synthetic research into 1 and its stereoisomers is very important in the study of structure–activity relationships of tetrahydrofuran lignans. The stereoselective introduction of three substituents onto a tetrahydrofuran ring is also an interesting topic. Some racemic syntheses of dihydrosesamin have been reported;³ however, there is no report on the synthesis of the optically active compound. This paper is the first report of a stereoselective synthesis of (+)-dihydrosesamin 2 which is the enantiomer of the natural product (Fig. 1).

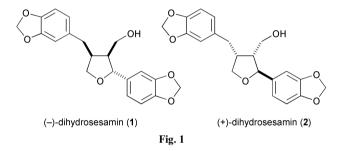
In this research, the *erythro*-selective aldol condensation using the potassium enolate of γ -butyrolactone⁴ was employed as a key reaction to obtain the stereochemistry at the 2- and 3-position of (+)-dihydrosesamin. It has been previously reported that aldol condensation of the potassium enolate of (3*R*)-3-(3,4-methylenedioxybenzyl)-4-butanolide⁵ **3** with some tri/dimethoxybenzaldehydes preferentially gave the *erythro* aldol product (Scheme 1).



Scheme 1 erythro-Selective aldol condensation.

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If the aldol condensation of the butanolide 3^5 with piperonal gave the *erythro* aldol product 4(2S,2'S) in high selectivity, this aldol product would be transformed into diol 5 which could be converted into the tetrahydrofuran-ring system 6 by S_N^2 cyclization with retention of the stereochemistry at the benzylic position. In this plan, the steric configuration at the 2- and 3position of the tetrahydrofuran ring in 6 would depend on the stereochemistry at the 2- and 2'-position of aldol product 4, respectively, and the steric configuration of the 3-position of the butanolide 3, whose preparation from L-glutamic acid has



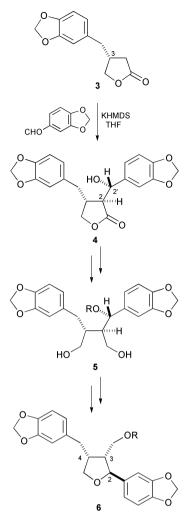
been reported,⁵ would be that of the 4-position of the tetrahydrofuran ring in 6 (Scheme 2).

Results and discussion

Aldol condensation of the butanolide 3^5 with piperonal using potassium bis(trimethylsilyl)amide as a base gave *erythro* aldol product 4 (2*S*,2'*S*) as a single isomer in 78% yield. Butanolide 3 was recovered to the extent of 14%. The coupling constant of 2-H with the proton of the benzyl substituent at C-2' (2.9 Hz) revealed the product to be the *erythro* isomer (2*S*,2'*S*).⁶ This coupling constant was due to the axial–equatorial relationship of 2-H and 2'-H in the six-membered ring formed by a hydrogen bond between the hydroxy group and the carbonyl group. In the case of the *threo* isomer, this coupling constant would be 6–9 Hz because of the resulting diaxial relationship.

After protection of the hydroxy group as a triethylsilyl (TES) group using triethylsilyl trifluoromethanesulfonate (TESOTf)⁷ and 2,6-dimethylpyridine (2,6-lutidine) (84%), the lactone 7 was reduced to diol 8 by lithium aluminium hydride in 100% yield. An $S_N 2$ cyclization was then adopted to produce the tetrahydrofuran ring. Thus diol 8 was converted into dimesyl derivative 9 by employing methanesulfonyl chloride and triethylamine, in 93% yield. This resulting unstable dimesyl diester was exposed to desilylation using tetra-n-butylammonium fluoride.8 In this stage, intramolecular $S_N 2$ cyclization occurred to give the (methylsulfonyloxymethyl)tetrahydrofuran 10 in 90% yield. This unstable monomesyl ester was treated with aq. sodium hydroxide in DMF to give (+)-dihydrosesamin 2 { $[a]_D^{20}$ +15.9, c 0.75 in pyridine; (-)-dihydrosesamin 1: $[a]_{D}^{20}$ -15.9, c 0.67 in pyridine¹}. An NOE experiment showed the correlation of the methylene protons of the hydroxymethyl group $(3'-H_2)$ with the benzylic protons on the 4-position (4'-H₂) and with

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Scheme 2 Synthetic plan of (+)-dihydrosesamin.

2-H (Scheme 3). The NMR and IR data of the synthesized (+)-dihydrosesamin 2 agreed with those of natural (-)-dihydrosesamin 1^{1,3}

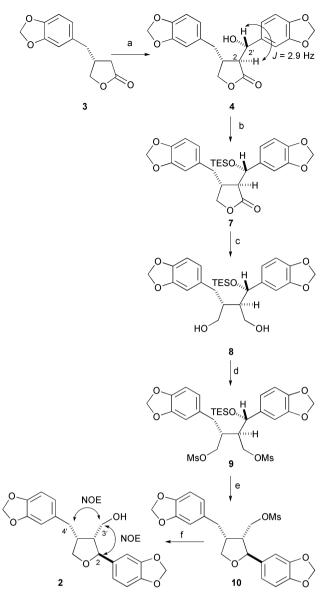
(+)-Dihydrosesamin 2 was therefore stereoselectively synthesized from the butanolide 3 in 6 steps in 37% overall yield. In this process, *erythro*-selective aldol condensation of butanolide 3 with piperonal was employed for stereoselective introduction of substituents.

Experimental

All melting-point (mp) data are uncorrected. NMR data were measured by a JNM-EX400 spectrometer. EIMS data were measured with an Hitachi M-80B and optical rotations were evaluated with an Horiba SEPA-200, $[a]_{\rm D}$ -values are in units of 10^{-1} deg cm² g⁻¹. The silica gel used was Wakogel C-300 (Wako, 200–300 mesh), and preparative TLC was conducted with Merck silica gel 60 F₂₅₄ (0.5 mm thickness, 20 × 20 cm). The numbering of compounds was changed to follow IUPAC nomenclature rules.

(2*S*,3*R*)-2-[(1*S*)-1-Hydroxy-1-(3,4-methylenedioxyphenyl)methyl]-3-(3,4-methylenedioxybenzyl)-4-butanolide 4

To a solution of KHMDS (0.5 M in toluene; 21.8 ml, 10.9 mmol) in THF (80 ml) was added a solution of the butanolide **3** (2.00 g, 9.08 mmol) in THF (10 ml) at -75 °C. After the solution had been stirred at -75 °C for 30 min, a solution of piperonal (1.57 g, 10.5 mmol) in THF (20 ml) was added. The reaction mixture was stirred at -75 °C for 1 h before the addition of saturated aq. NH₄Cl and EtOAc. The organic solution was separated, washed with brine, dried (Na₂SO₄),



Scheme 3 Synthesis of (+)-dihydrosesamin 2. Reagents and conditions (yields): (a) piperonal, KHMDS, THF, -75 °C, 1 h (78%); (b) TESOTf, 2,6-lutidine, CH₂Cl₂, rt, 1 h (84%); (c) LiAlH₄, THF, 0 °C, 1 h (100%); (d) MsCl, Et₃N, CH₂Cl₂, rt, 1 h (93%); (e) *n*-Bu₄NF, THF, 0 °C, 1.5 h (90%); (f) 1 M aq. NaOH, DMF, 120 °C, 16 h (68%).

and evaporated. The residue was purified by silica gel column chromatography (10% EtOAc-benzene) to give erythro-aldol product 4 (2.62 g, 78%) as a colorless oil, $[a]_{D}^{20}$ -44.6 (c 0.92, CHCl₃); v_{max}(CHCl₃)/cm⁻¹ 3500, 2899, 1755, 1611, 1505, 1445, 1042; $\delta_{\rm H}$ (CDCl₃) 2.27 (1H, dd, J 13.7, 7.8 Hz, ArCH₂), 2.42 (1H, dd, J 13.7, 8.1 Hz, ArCH₂), 2.58 (1H, dd, J 5.9, 2.9 Hz, 2-H), 2.79 (1H, m, 3-H), 2.82 (1H, s, OH), 3.94 (1H, dd, J 8.8, 5.9 Hz, 4-H), 4.32 (1H, dd, J 8.8, 8.3 Hz, 4-H), 5.24 (1H, br s, by D₂O-exchange d, J 2.9 Hz, ArCHOH), 5.89 (1H, d, J 1.5 Hz, OCHHO), 5.92 (1H, d, J 1.5 Hz, OCHHO), 5.95 (1H, d, J 1.5 Hz, OCHHO), 5.96 (1H, d, J 1.5 Hz, OCHHO), 6.30-6.34 (2H, m, ArH), 6.61 (1H, d, J 7.8 Hz, ArH), 6.72 (3H, s, ArH); $\delta_{\rm C}({\rm CDCl}_3)$ 36.3, 39.4, 52.8, 71.8, 72.7, 100.9, 101.1, 105.8, 108.1, 108.6, 118.3, 121.5, 131.5, 134.9, 146.1, 146.9, 147.6, 147.8, 178.3; m/z (EI, 20 eV) 370 (M⁺, 11%), 220 (23), 135 (100) [Found (HRMS): M^+ , 370.1048. $C_{20}H_{18}O_7$ requires M, 370.1051].

(2*S*,3*R*)-3-(3,4-Methylenedioxybenzyl)-2-[(*S*)-(3,4-methylenedioxyphenyl)(triethylsilyloxy)methyl]-4-butanolide 7

To an ice-cooled solution of aldol product 4 (2.30 g, 6.21 mmol) and 2,6-lutidine (1.81 ml, 15.5 mmol) in CH_2Cl_2

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(10 ml) was added TESOTf (1.62 ml, 7.16 mmol). The reaction solution was stirred at room temperature for 1 h before addition of saturated aq. NaHCO₃. The organic solution was separated, washed with brine, and dried (Na₂SO₄). After evaporation, the residue was purified by silica gel column chromatography (40% EtOAc-hexane) to give silvl ether 7 (2.53 g, 84%) as colorless crystals, mp 114-115 °C (30% EtOAchexane), $[a]_{D}^{20}$ -54.3 (c 0.92, CHCl₃); v_{max} (CHCl₃)/cm⁻¹ 2880, 1763, 1611, 1505, 1443, 1042; δ_H(CDCl₃) 0.56 (6H, q, J 7.8 Hz, SiCH₂CH₃), 0.88 (9H, t, J 7.8 Hz, SiCH₂CH₃), 2.23 (1H, dd, J 13.7, 9.3 Hz, ArCH₂), 2.41 (1H, dd, J 3.9, 2.0 Hz, 2-H), 2.49 (1H, dd, J 13.7, 7.3 Hz, ArCH₂), 2.81 (1H, m, 3-H), 3.98 (1H, dd, J 8.5, 3.7 Hz, 4-H), 4.39 (1H, dd, J 8.5, 7.8 Hz, 4-H), 5.27 (1H, d, J 2.0 Hz, ArCHOH), 5.89 (1H, d, J 1.5 Hz, OCHHO), 5.91 (1H, d, J 1.5 Hz, OCHHO), 5.95 (1H, d, J 1.5 Hz, OCHHO), 5.97 (1H, d, J 1.5 Hz, OCHHO), 6.21 (1H, d, J 1.5 Hz, ArH), 6.27 (1H, dd, J 7.8, 1.5 Hz, ArH), 6.56 (1H, d, J 7.8 Hz, ArH), 6.64–6.65 (3H, m, ArH); δ_c(CDCl₃) 4.6, 6.7, 35.9, 39.9, 54.3, 73.2, 73.3, 100.9, 101.1, 105.7, 107.8, 107.9, 108.6, 118.2, 121.7, 131.6, 135.9, 146.0, 146.7, 147.5, 178.1; m/z (EI, 20 eV) 484 (M⁺, 5%), 455 (22), 265 (100), 135 (33) (Found: C, 64.31; H, 6.52. C₂₆H₃₂O₇Si requires C, 64.44; H, 6.66%).

(2*R*,3*R*)-2-(3,4-Methylenedioxybenzyl)-3-[(*S*)-(3,4-methylenedioxyphenyl)(triethylsilyloxy)methyl]butane-1,4-diol 8

To a suspension of LiAlH₄ (0.13 g, 3.43 mmol) in THF (5 ml) was added a solution of lactone 7 (1.10 g, 2.27 mmol) in THF (10 ml) at 0 °C. After the reaction mixture had been stirred at room temperature for 1 h, saturated aq. MgSO₄ and K₂CO₃ were added. The mixture was stirred at room temperature for 30 min before filtration. The filtrate was concentrated. The residue was subjected to silica gel column chromatography (EtOAc-hexane 1:2) to give diol 8 (1.11 g, 100%) as a colorless oil, $[a]_{D}^{20}$ -60.0 (c 0.85, CHCl₃); v_{max} (CHCl₃)/cm⁻¹ 3425, 2880, 1609, 1505, 1443, 1042, 939; $\delta_{\rm H}$ (CDCl₃) 0.52 (6H, q, J 7.8 Hz, SiCH₂CH₃), 0.87 (9H, t, J 7.8 Hz, SiCH₂CH₃), 1.79 (1H, m), 2.29 (1H, m), 2.38 (1H, dd, J 13.7, 8.3 Hz, ArCH₂), 2.50–2.70 (1H, br, OH), 2.66 (1H, dd, J 13.7, 7.1 Hz, ArCH₂), 3.46 (1H, dd, J 11.0, 7.1 Hz, CH₂OH), 3.53 (1H, dd, J 11.0, 4.6 Hz, CH₂OH), 3.65 (1H, dd, J 11.1, 2.5 Hz, CH₂OH), 3.77 (1H, dd, J 11.1, 5.6 Hz, CH₂OH), 3.80–4.00 (1H, br, OH), 4.92 (1H, d, J 6.3 Hz, ArCHOH), 5.90 (2H, s, OCH₂O), 5.95 (2H, d, J 3.4 Hz, OCH₂O), 6.44 (1H, s, ArH), 6.46 (1H, d, J 7.8 Hz, ArH), 6.64 (1H, d, J 7.8 Hz, ArH), 6.68–6.70 (3H, m, ArH); $\delta_{\rm C}({\rm CDCl}_3)$ 4.7, 6.7, 36.8, 39.4, 50.0, 59.5, 61.5, 73.7, 100.7, 100.9, 106.9, 107.6, 107.9, 109.2, 119.7, 121.8, 134.0, 137.2, 145.6, 146.5, 147.4, 147.5; m/z (EI, 20 eV) 488 (M⁺, 0.2%), 265 (100), 135 (62) (Found: C, 64.26; H, 6.54. C₂₆H₃₆O₇Si requires C, 63.91; H, 7.43%).

(2*S*,3*R*,4*R*)-3-Hydroxymethyl-4-(3,4-methylenedioxybenzyl)-2-(3,4-methylenedioxyphenyl)tetrahydrofuran [(+)-dihydrosesamin] 2

To an ice-cooled solution of diol **8** (0.55 g, 1.13 mmol) and Et₃N (0.33 ml, 2.37 mmol) in CH₂Cl₂ (10 ml) was added MsCl (0.19 ml, 2.45 mmol). After the reaction solution had been stirred at room temperature for 1 h, saturated aq. NaHCO₃ and CH₂Cl₂ were added. The organic solution was separated, washed with brine, and dried (Na₂SO₄). Evaporation and silica gel column chromatography (EtOAc–hexane 1 : 2) gave unstable dimesyl derivative **9** (0.68 g, 93%) as a colorless oil, $\delta_{\rm H}$ (CDCl₃) 0.53 (6H, q, *J* 7.8 Hz), 0.89 (9H, t, *J* 7.8 Hz), 2.08 (1H, m), 2.42 (1H, m), 2.53 (1H, dd, *J* 13.7, 9.3 Hz), 2.79 (1H, dd, *J* 13.7, 6.1 Hz), 3.00 (3H, s, OMs), 3.01 (3H, s, OMs), 4.23 (2H, d, *J* 6.4 Hz), 4.86 (1H, d, *J* 4.9 Hz), 5.92 (2H, s), 5.97 (2H, d, *J* 2.4 Hz), 6.39 (1H, s), 6.45 (1H, d, *J* 8.3 Hz), 6.63 (1H, d, *J* 8.3

Hz), 6.61 (1H, s), 6.67 (1H, d, *J* 8.3 Hz), 6.70 (1H, d, *J* 8.3 Hz); $\delta_{\rm C}$ (CDCl₃) 4.7, 6.8, 35.7, 37.2, 37.4, 46.3, 67.7, 70.7, 73.4, 100.9, 101.1, 106.4, 107.8, 108.0, 108.9, 119.4, 121.9, 132.2, 136.0, 146.0, 146.9, 147.6.

To an ice-cooled solution of this dimesyl compound (0.68 g, 1.05 mmol) in THF (10 ml) was added n-Bu₄NF (1 M in THF; 1.05 ml, 1.05 mmol). After the reaction solution had been stirred at 0 $^\circ C$ for 1.5 h, saturated aq. $\rm NH_4Cl$ and EtOAc were added. The organic solution was separated, washed with brine, and dried (Na₂SO₄). Evaporation and silica gel column chromatography (EtOAc-hexane 1 : 2) gave the unstable (methylsulfonyloxymethyl)tetrahydrofuran 10 (0.41 g, 90%) as a colorless oil, $\delta_{\rm H}$ (CDCl₃) 2.49–2.58 (2H, m), 2.74 (1H, m), 2.85 (1H, dd, J 13.4, 5.1 Hz), 3.01 (3H, s, OMs), 3.71 (1H, dd, J 8.8, 7.1 Hz), 4.07 (1H, dd, J 8.8, 6.8 Hz), 4.29 (1H, dd, J 10.0, 8.6 Hz), 4.45 (1H, dd, J 10.0, 8.6 Hz), 4.80 (1H, d, J 5.9 Hz), 5.94 (2H, s), 5.95 (2H, s), 6.61-6.69 (2H, m), 6.73-6.81 (4H, m); $\delta_{\rm C}({\rm CDCl}_3)$ 33.0, 37.5, 42.1, 49.5, 67.2, 72.6, 82.4, 100.9, 101.1, 106.1, 108.1, 108.3, 108.8, 119.1, 121.4, 133.2, 135.8, 147.2, 147.9.

A reaction solution of the (methylsulfonyloxymethyl)tetrahydrofuran 10 (0.41 g, 0.98 mmol) in DMF (0.5 ml) and 1 M aq. NaOH (0.5 ml) was stirred at 120 °C for 16 h before additions of water and EtOAc. The organic solution was separated, washed with brine, dried (Na₂SO₄), and evaporated. The residue was applied to silica gel TLC plates (benzene-EtOAc 5 : 1) to give (+)-dihydrosesamin 2 (0.24 g, 68%) as colorless crystals, mp 97–98 °C; $[a]_{\rm D}^{20}$ +15.9 (*c* 0.75, pyridine); $v_{\rm max}$ (CHCl₃)/cm⁻¹ 3437, 2888, 1611, 1505, 1445, 1042, 938; $\delta_{\rm H}(\rm CDCl_3)$ 1.60 (1H, br s, OH), 2.35 (1H, m, 3-H), 2.53 (1H, dd, J 13.4, 10.5 Hz, ArCH₂), 2.70 (1H, m, 4-H), 2.87 (1H, dd, J 13.4, 5.1 Hz, ArCH₂), 3.72 (1H, dd, J 8.3, 6.4 Hz, 5-H), 3.76 (1H, dd, J 10.7, 6.8 Hz, CH₂OH), 3.89 (1H, dd, J 10.7, 6.8 Hz, CH₂OH), 4.04 (1H, dd, J 8.3, 6.6 Hz, 5-H), 4.79 (1H, d, J 6.4 Hz, 2-H), 5.93 (2H, s, OCH₂O), 5.94 (2H, s, OCH₂O), 6.64 (1H, d, J7.8 Hz, ArH), 6.68 (1H, s, ArH), 6.73 (1H, d, J7.8 Hz, ArH), 6.76 (2H, s, ArH), 6.83 (1H, s, ArH); δ_c(CDCl₃) 33.2 (ArCH₂), 42.3 (4-C), 52.6 (3-C), 60.9 (CH₂OH), 72.9 (5-C), 82.9 (2-C), 100.9 (OCH₂O), 101.0 (OCH₂O), 106.3, 108.0, 108.3, 108.9, 119.0, 121.4, 134.1, 137.0, 145.9, 146.9, 147.7, 147.8; m/z (EI, 20 eV) 356 (M⁺, 65%), 135 (100) [Found (HRMS): M⁺, 356.1252. C₂₀H₂₀O₆ requires *M*, 356.1258].

Acknowledgements

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